

constant	symbol	value
Avogadro's number	N_A	6.02214×10^{23}
gas constant	R	$0.0820574 \text{ L}\cdot\text{atm}\cdot\text{mol}^{-1}\cdot\text{K}^{-1}$ $8.31446 \text{ J}\cdot\text{mol}^{-1}\cdot\text{K}^{-1}$
speed of light in a vacuum	c	$299792458 \text{ m}\cdot\text{s}^{-1}$ (exact)
Planck constant	h	$6.62607 \times 10^{-34} \text{ J}\cdot\text{s}$
Rydberg energy	R_y	$2.17987 \times 10^{-18} \text{ J}$
Boltzmann constant	k_B	$1.38065 \times 10^{-23} \text{ J}\cdot\text{K}^{-1}$
elementary charge	e	$1.60218 \times 10^{-19} \text{ C}$
mass of electron	m_e	$9.10938 \times 10^{-31} \text{ kg}$
atomic mass unit	u	$1.66054 \times 10^{-27} \text{ kg}$
standard acceleration due to Earth's gravity	g	$9.80665 \text{ m}\cdot\text{s}^{-2}$ (exact)
ion product of water	K_w	$1.01158 \times 10^{-14} M^2$
Faraday constant	F	$96485.3 \text{ C}\cdot\text{mol}^{-1}$
Coulomb constant	k	$8.98755 \times 10^9 \text{ N}\cdot\text{m}^2\cdot\text{C}^{-2}$
permittivity of free space	ϵ_0	$8.85419 \times 10^{-12} \text{ N}^{-1}\cdot\text{m}^{-2}\cdot\text{C}^2$

name	chemical formula	melting point (T_m) °C	boiling point (T_b) °C	molal freezing point depression constant (K_f) °C·kg·mol ⁻¹	molal boiling point elevation constant (K_b) °C·kg·mol ⁻¹
acetic acid	CH ₃ CO ₂ H	17.0	117.9	3.63	3.22
benzene	C ₆ H ₆	5.538	80.08	5.07	2.64
benzotrile	C ₆ H ₅ CN	-12.82	191	5.35	--
carbon disulfide	CS ₂	-111.7	46.2	--	2.42
carbon tetrachloride	CCl ₄	-22.8	76.7	--	5.26
cyclohexane	C ₆ H ₁₂	6.7	80.7	20.8	2.92
dibenzyl ether	(C ₆ H ₅ CH ₂) ₂ O	1.8	298	6.17	--
diethyl ether	CH ₃ CH ₂ OCH ₂ CH ₃	-116.22	34.4	--	2.2
ethanol	CH ₃ CH ₂ OH	-114.14	78.24	--	1.23
ethylene glycol	(CH ₂ OH) ₂	-13.0	197.5	3.11	2.26
formamide	NH ₂ COH	2.57	217	4.25	--
hexane	C ₆ H ₁₄	-95.27	68.72	--	2.9
methanol	CH ₃ OH	-97.5	64.5	--	2.9
naphthalene	C ₁₀ H ₈	80.22	218	7.45	--
water	H ₂ O	0	99.974	1.86	0.513